Scheikunde hoofdstuk 1  
Tabel 1 , 2 en 3

Tabel 1 namen en formules van 2-atomige niet-ontleedbare stoffen

|  |  |
| --- | --- |
| Stof | Formule |
| Waterstof | H2 (g) |
| Stikstof | N2 (g) |
| Zuurstof | O2 (g) |
| Fluor | F2 (g) |
| Chloor | Cl2 (g) |
| Broom | Br2 (l) |
| Jood | I2 (g) |

Tabel 2 namen en formules van ontleedbare stoffen

|  |  |
| --- | --- |
| Naam | Formule |
| Water | H2O (l) |
| Waterstofperoxide | H2O2 (l) |
| Ammoniak | NH3 (g) |
| Koolstofomono-oxide | CO (g) |
| Koolstofdioxide | CO2 (g) |
| Zwaveldioxide | SO2 (g) |
| Zwaveltrioxide | SO3 (g) |
| Natriumchoride | NaCl (s) |
| Alcohol | C2H6O (l) |
| Glucose | C6H12O6 (s) |

Tabel 3 elementen en hun symbool

|  |  |  |  |
| --- | --- | --- | --- |
| Metaal | Symbool | Niet-metaal | Symbool |
| Zilver | Ag | Argon | Ar |
| Aluminium | Al | Broom | Br |
| Goud | Au | Koolstof | C |
| Barium | Ba | Chloor | Cl |
| Calcium | Ca | Fluor | F |
| Cadmium | Cd | Waterstof | H |
| Chroom | Cr | Helium | He |
| Koper | Cu | Jood | I |
| Ijzer | Fe | Stikstof | N |
| Kwik | Hg | Neon | Ne |
| Kaluim | K | Zuurstof | O |
| Magnesium | Mg | Fosfor | P |
| Natrium | Na | Zwavel | S |
| Nikkel | Ni | Silicium | Si |
| Lood | Pb |  |  |
| Platina | Pt |  |  |
| Tin | Sn |  |  |
| Zink | Zn |  |  |